# Substituted 2-[(4-Aminomethyl)phenoxy]-2-methylpropionic Acid PPAR $\alpha$ Agonists. 1. Discovery of a Novel Series of Potent HDLc Raising Agents 

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The peroxisome proliferator activated receptors PPAR $\alpha, \operatorname{PPAR} \gamma$, and PPAR $\delta$ are ligand-activated transcription factors that play a key role in lipid homeostasis. The fibrates raise circulating levels of high-density lipoprotein cholesterol and lower levels of triglycerides in part through their activity as PPAR $\alpha$ agonists; however, the low potency and restricted selectivity of the fibrates may limit their efficacy, and it would be desirable to develop more potent and selective PPAR $\alpha$ agonists. Modification of the selective PPAR $\delta$ agonist 1 (GW501516) so as to incorporate the 2-aryl-2-methylpropionic acid group of the fibrates led to a marked shift in potency and selectivity toward PPAR $\alpha$ agonism. Optimization of the series gave 25a, which shows $\mathrm{EC}_{50}=4 \mathrm{nM}$ on PPAR $\alpha$ and at least 500 -fold selectivity versus PPAR $\delta$ and PPAR $\gamma$. Compound 25a (GW590735) has been progressed to clinical trials for the treatment of diseases of lipid imbalance.

## Introduction

Elevated circulating levels of low-density lipoprotein cholesterol (LDLc) constitutes a major risk factor for coronary artery disease, and there exists a wealth of clinical data supporting the use of the LDLc-lowering statins in an increasingly wide range of patients. ${ }^{1}$ More recently, the roles of both low circulating levels of high-density lipoprotein cholesterol (HDLc) and high circulating levels of triglycerides (TG) as cardiovascular disease risk factors have come into focus, and there is a growing level of confidence in the potential of drugs that lower TG or raise HDLc to play an important part in future cardiovascular drug therapy. ${ }^{2}$

The principal agents in current use for raising HDLc and lowering TG are the fibrates and nicotinic acid. The fibrates give increases of up to $20 \%$ in HDLc and decreases in TG of up to $40 \% ;{ }^{2,3}$ nicotinic acid shows similar efficacy on HDLc with perhaps a slightly less pronounced effect on TG. It would be highly desirable to discover drugs that exert greater effects on HDLc and TG levels.

While the molecular target of nicotinic acid has only recently been identified, ${ }^{4}$ the fibrates have been known for several years

[^0]Table 1. Human PPAR $\alpha$ Agonist Potencies of Fibrates ${ }^{a}$
Drug
${ }^{a}$ Data were taken from ref 10 and were generated using the PPARGAL4 transactivation assay.
to act as agonists at the peroxisome proliferator activated receptor alpha (PPAR $\alpha$ ). ${ }^{5,6}$ The PPARs comprise a family of ligand-activated transcription factors that play a key role in lipid homeostasis. There are three members of the family: PPAR $\alpha$, $\operatorname{PPAR} \delta$ (or $\beta$ ), and PPAR $\gamma .{ }^{7} \operatorname{PPAR} \alpha$ is highly expressed in the liver, heart, and muscle and includes a range of fatty acids among its natural ligands. In man, activation of PPAR $\alpha$ results in increased clearance of TG-rich very low-density lipoprotein (VLDL) via a reduction in plasma levels of ApoCIII ${ }^{8}$ and in upregulation of ApoA1, the principal lipoprotein component of HDL. ${ }^{9}$ As a consequence, the fibrates lower TG and raise HDLc levels. The fibrates are only weakly potent PPAR $\alpha$ agonists, ${ }^{10}$ as shown in Table 1. It has been proposed that more potent and subtype-selective PPAR $\alpha$ activators might offer enhanced specificity and reduced side effects compared with that of existing fibrates, ${ }^{11}$ and several groups have reported the

Scheme 1


Scheme $\mathbf{2}^{a}$


[^1] reflux.

## Scheme $3^{a}$


${ }^{a}$ Reagents and conditions: (a) 4-methoxybenzyltriphenylphosphonium chloride, $\mathrm{NaH}\left(60 \%\right.$ in mineral oil), $\mathrm{CH}_{2} \mathrm{Cl}{ }_{2}$; (b) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH} / \mathrm{AcOH}, \mathrm{H} 2$; (c) $\mathrm{BBr}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2},-40^{\circ} \mathrm{C}$; (d) 2-trichloromethyl-2-propanol, acetone, NaOH (conc), rt.
discovery of potent PPAR $\alpha$ agonists and their effects in animal models of dyslipidemia. ${ }^{12}$

We describe below how, starting from the recently published ${ }^{13}$ selective PPAR $\delta$ agonist $\mathbf{1}$ (GW501516) which showed weak potency on PPARa, we have first incorporated the classical fibrate "head group" to improve selectivity for PPAR $\alpha$ and subsequently optimized potency and selectivity. We have thus developed a new series of highly potent and selective PPAR $\alpha$ agonists, one of which, 25a, is currently in clinical trials for the treatment of dyslipidemia (Scheme 1).

## Chemistry

Compound 10 was prepared as shown in Scheme 2. Condensation of 4-trifluoromethylthiobenzamide 2a with ethyl-2chloroacetoacetate gave the thiazole 3a, which was reduced with
$\mathrm{LiAlH}_{4}$ and oxidized with PCC to give the aldehyde 5. Compound 7 was prepared in two steps from ethyl 2-methylphenoxyacetate via reaction with formaldehyde and HCl to give 6 followed by substitution with triphenylphosphine to give 7. Wittig condensation of 5 with 7 gave the alkene 8 , which was hydrogenated over $10 \% \mathrm{Pd}$ on carbon to give 9. Hydrolysis using NaOH in dioxane furnished compound $\mathbf{1 0}$.

Compound 14 was synthesized according to the procedure depicted in Scheme 3. Wittig condensation of 5 with 4 -methoxybenzyltriphenylphosphonium chloride gave 11, which was hydrogenated over $10 \% \mathrm{Pd}$ on carbon to give $\mathbf{1 2}$. Demethylation using boron tribromide gave 13, and coupling with 2-trichloro-methyl-2-propanol and hydrolysis afforded 14.

Alkylation of 4-nitrophenol followed by hydrogenolysis gave 16, which was coupled with the acid 17a, prepared by hydrolysis

Scheme $4^{a}$

${ }^{a}$ Reagents and conditions: (a) ethyl 2-bromo-2-methylpropionate, DMF, $\mathrm{K}_{2} \mathrm{CO}_{3}, 40^{\circ} \mathrm{C}$; (b) $5 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH}, 30^{\circ} \mathrm{C}, \mathrm{H}_{2}$; (c) NaOH (aq), EtOH , reflux; (d) HOBT, EDC, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{DMF}$, rt; (e) $\mathrm{NaOH}(1 \mathrm{~N}), \mathrm{EtOH}, 40^{\circ}$.

Scheme $5^{a}$

${ }^{a}$ Reagents and conditions: (a) NaH ( $60 \%$ in mineral oil), DMF, ethyl-2-bromo-2-methylpropionate, rt then reflux; (b) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{AcOH} / \mathrm{EtOH}$, $\mathrm{H}_{2}$; (c) 5, $\mathrm{NaBH}(\mathrm{OAc})_{3}$, dichloroethane/AcOH, rt; (d) $\mathrm{NaOH}(1 \mathrm{~N}), \mathrm{EtOH}$, $40^{\circ}$.

Scheme $\mathbf{6}^{a}$

${ }^{a}$ Reagents and conditions: (a) ethyl-2-chloroacetoacetate, EtOH, reflux; (b) $\mathrm{NaOH}(1 \mathrm{~N}), \mathrm{EtOH}$, reflux.
of ester 3a, to give 18. Hydrolysis using NaOH in EtOH gave 19, as shown in Scheme 4.

Compound 23 was prepared from 4-cyanophenol via alkylation and hydrogenation to give 21, which was reacted with aldehyde 5 under reductive amination conditions. The standard hydrolysis gave 23 (Scheme 5).

Thiazole carboxylic acids $\mathbf{1 7 b} \mathbf{-} \mathbf{j}$ were prepared in the same manner as $\mathbf{1 7 a}$, as indicated in Scheme 6. The acids $\mathbf{1 7 a}-\mathbf{j}$ were then elaborated to the amides $\mathbf{2 5 a} \mathbf{-} \mathbf{j}$ via coupling using either HOBT/EDC/Et ${ }_{3} \mathrm{~N}$ or $\mathrm{SOCl}_{2}$, followed by hydrolysis (Scheme 7).

N -Methylamide 27 was obtained by methylation of 24a using NaH and methyl iodide followed by hydrolysis, as shown in Scheme 8.

The reversed amide 34 was prepared as shown in Scheme 9. Bromination of 4 using $\mathrm{PBr}_{3}$ gave 28. Substitution by potassium phthalimide gave 29 , which was cleaved by hydrazine to give the amine 30. Compound 30 was coupled with 32, in turn prepared by alkylation of 4-hydroxybenzaldehyde followed by oxidation using sodium chlorite and sodium hydrogen phosphate, giving 33, which was then hydrolyzed to give 34.

The homologated amide 37 was synthesized via HOBT coupling of 17 a with 4-hydroxyphenethylamine followed by alkylation and hydrolysis, as shown in Scheme 10.

Compound 25k was prepared by hydrogenolysis of $\mathbf{2 4} \mathbf{e}$ followed by hydrolysis (Scheme 11).

The 4-methanesulfonyloxy thiazole carboxylic ester $\mathbf{3 1}$ was prepared from 4-hydroxythiobenzamide $\mathbf{2 k}$ using the procedure described above, followed by mesylation using methane sulfonyl chloride and triethylamine. LiOH mediated hydrolysis gave the carboxylic acid $\mathbf{1 7 k}$, which was alkylated to give ester $\mathbf{2 4 l}$. Hydrolysis under mild conditions ( 1 equiv of $\mathrm{LiOH}, \mathrm{rt}$ ) of 241 gave product 251, whereas more vigorous hydrolysis of 241 ( $\mathrm{NaOH}, \mathrm{THF} / \mathrm{EtOH}, 70^{\circ}$ ) gave the phenol 25m (Scheme 12).

The regioisomeric thiazole 41 was prepared in a similar manner to 25a via the carboxylic acid 39, obtained from condensation of $\mathbf{2 a}$ in refluxing ethanol with the intermediate obtained from the reaction of 2-ketobutyric acid with bromine to give 38, followed by hydrolysis. Coupling of 39 with 21 followed by standard hydrolysis gave 41, as depicted in Scheme 13.

The trifluoromethyl-substituted thiazole $\mathbf{4 5}$ was prepared using a similar route to that used for 25a, substituting ethyl-2-chloro-4,4,4-trifluoroacetate for the methyl analogue (Scheme 14).

## Results and Discussion

The recently published $\mathbf{1}$ is a potent agonist at $\operatorname{PPAR} \delta$ with around 1000 -fold selectivity with respect to PPAR $\alpha$, as measured using cell-based transient-transfection assays ${ }^{14}$ (Table 2).

Replacement of the thiomethylene chain of $\mathbf{1}$ by ethylene gave 10, which showed equally low potency against PPAR $\alpha$. PPAR $\delta$ potency was somewhat decreased $\left(\mathrm{EC}_{50}=4 \mathrm{nM}\right)$ relative to $\mathbf{1}$, while $\operatorname{PPAR} \gamma$ activity remained very weak. Encouragingly however, replacement of the o-methyl phenoxyacetate group by the classical "fibrate head group" shared by all of the drugs shown in Table 1 gave 14, which regained potency against $\operatorname{PPAR} \alpha\left(\mathrm{EC}_{50}=210 \mathrm{nM}\right)$.

We further explored the effect of modifying the chain linking the head group to the biaryl unit (Table 3). Replacement of the ethylene of $\mathbf{1 4}$ by an amide link gave $\mathbf{1 9}$, which showed almost complete loss of activity. However, increasing the length of the linking chain proved beneficial for PPAR $\alpha$ potency. The

Scheme $7^{a}$

${ }^{a}$ Reagents and conditions: Method $\mathrm{A}, \mathrm{SOCl}_{2}, \mathbf{2 1}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; Method B, 21, HOBT, EDC, $\mathrm{Et}_{3} \mathrm{~N}$, DMF, rt; (a) $\mathrm{NaOH}(1 \mathrm{~N})$, THF, reflux.

Scheme $\mathbf{8}^{a}$


24a
a

$\mathrm{b} \downarrow^{26}: \mathrm{R}=\mathrm{Et}$
27 : $\mathrm{R}=\mathrm{H}$
${ }^{a}$ Reagents and conditions: (a) NaH ( $60 \%$ in mineral oil), MeI, DMF, $40^{\circ} \mathrm{C}$; (b) $\mathrm{NaOH}(1 \mathrm{~N}), \mathrm{EtOH}$, reflux.
methyleneaminomethylene derivative 23 gained 3-fold in potency, and the amide 25a showed a 50 -fold increase in potency. N -Methylation (27) of the amide 25a was detrimental to PPAR $\alpha$ potency, and the inverse amide $\mathbf{3 4}$ and the chain-extended amide 37 were also less potent.

Replacement of the terminal trifluoromethyl substituent by a range of groups showed that substitution in this position was largely well tolerated and that potency correlated well with the lipophilicity of the substituent (Table 4). Thus the tert-butyl, trifluoromethyl, and trifluoromethoxy groups gave the highest potency, while fluoro, hydrogen, and amino substituents gave a progressive decrease in potency. All compounds demonstrated considerable selectivity for PPAR $\alpha$.

The $m$-trifluoromethyl analogue $\mathbf{2 5 i}$ was equipotent with $\mathbf{2 5 a}$ versus PPAR $\alpha$, whereas the $o$-trifluoromethyl analogue $\mathbf{2 5 j}$ lost both PPAR $\alpha$ potency and selectivity versus PPAR $\delta$. The p-hydroxy $\mathbf{2 5 m}$ and $p$-methanesulfonyloxy $\mathbf{2 5 1}$ substituted compounds were inactive.

We next investigated the effect of modifying the central heterocycle fragment (Table 5). The thiazole regioisomer 41 showed a 500 -fold decrease in PPAR $\alpha$ potency compared with 25a, whereas PPAR $\delta$ potency increased. The trifluoromethyl analogue $\mathbf{4 5}$ showed similar PPAR $\alpha$ agonist potency to that of 25a but was less selective versus PPAR $\gamma$.

Binding Mode of 25a in the PPAR $\alpha$ Ligand Binding Domain. To determine with precision the binding mode of this new series of ligands, 25a was cocrystallized with PPAR $\alpha$. As
shown in Figure 1, 25a fits well into the PPAR $\alpha$ binding site. The carboxylate moiety of the head group forms four hydrogen bonds with Ser-280, Tyr-314, Tyr-464, and His-440, as expected from the previous PPAR $\alpha$ X-ray crystal structures of GW409544 ${ }^{15}$ and AZ242. ${ }^{16}$ The gem-dimethyl substituents are directed into a lipophilic pocket bounded by Phe-273, Gln-277, Val-444, and Leu-456, a region at the top end of the so-called "benzophenone" pocket. ${ }^{17}$ These residues are conserved in $\operatorname{PPAR} \gamma$ and PPAR $\delta$ except for Val-444, which is leucine in $\operatorname{PPAR} \gamma$ and methionine in PPAR $\delta$. This side chain is smallest and most able to accommodate the fibrate headgroup in PPAR $\alpha$ and is largest and least able to accommodate the fibrate headgroup, in PPAR $\delta . .^{14,18}$ This might explain why 14 gains potency versus PPAR $\alpha$ and PPAR $\gamma$ while maintaining a similar level of PPAR $\delta$ potency compared with $\mathbf{1 0}$.

The amide group of 25a adopts a trans-conformation in which the carbonyl oxygen is directed toward Thr-279 and nearby water molecules. Although the carbonyl oxygen of 25a lies near Thr-279, the threonine side chain appears to adopt a conformation that turns its hydroxyl group away from the carbonyl oxygen. Instead of hydrogen bonding with this threonine, the carbonyl oxygen hydrogen bonds indirectly with Ser-280 and Thr-283 via two water molecules (Figure 2a). Thr-279 is hydrogen bonded to the backbone NH of Ala-333 via a third water molecule and oriented toward the sulfur of the thiazole in a manner likely to give a favorable interaction, even though sulfur does not form strong hydrogen bonds.

The amide NH is positioned close to the methyl group of the thiazole and surrounded by three sulfur-containing amino acids, Cys-276, Met-355, and Met-330, with the sulfur atoms directed toward the NH proton at distances of $4.38 \AA, 2.92 \AA$, and 3.33 $\AA$, respectively (Figure 2 b ). This relatively polar environment stabilizes the partial positive charge of the NH , thereby favoring the observed conformation of 25a. Overall, the linking amide group of 25a appears to be well stabilized by the protein environment with the carbonyl surrounded by a very strong hydrogen bond network and the NH by a triad of polar atoms. This remarkable environment might explain the potency and selectivity of the amide linker compound in the PPAR $\alpha$ pocket, which is more lipophilic and less solvent-exposed than the corresponding pockets of either PPAR $\gamma$ or PPAR $\delta .{ }^{15}$

The precise binding network around the amide linker might also account for the observation that any modification resulted in a decrease in PPAR $\alpha$ potency. N -Methylation in 27 is not well tolerated due to the proximity of the sulfur atoms, and the reverse amide of $\mathbf{3 4}$ binds less favorably because its carbonyl

Scheme $\mathbf{9}^{a}$


$e \downarrow^{31: R}=\mathrm{CHO}, \begin{aligned} & \text { R } \\ & 32: R=\mathrm{CO}_{2} \mathrm{H}\end{aligned}$


$g f^{33:} \begin{array}{r}R \\ 34: R t \\ \end{array}$
${ }^{a}$ Reagents and conditions: (a) $\mathrm{PBr}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}$; (b) potassium phthalimidate, DMF, $70{ }^{\circ} \mathrm{C}$; (c) hydrazine hydrate, $\mathrm{EtOH}, 90^{\circ} \mathrm{C}$; (d) ethyl-2-bromo-2-methylpropionate, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{DMF}, 40^{\circ} \mathrm{C}$; (e) $\mathrm{NaClO}_{2}, \mathrm{NaHPO}_{3}, \mathrm{H}_{2} \mathrm{O}, t-\mathrm{BuOH}$, rt; (f) $\mathrm{SOCl}_{2}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; (g) NaOH (1 N), THF, reflux.

## Scheme 10 ${ }^{a}$


${ }^{a}$ Reagents and conditions: (a) 17a, HOBT, EDC, DMF, $\mathrm{Et}_{3} \mathrm{~N}$, rt; (b) ethyl-2-bromo-2-methylpropionate, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, $80^{\circ} \mathrm{C}$; (c) $\mathrm{NaOH}(1 \mathrm{~N})$, THF, reflux.

## Scheme $11^{a}$


${ }^{a}$ Reagents and conditions: (a) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH}, \mathrm{rt}$; (b) $\mathrm{NaOH}(1 \mathrm{~N})$, THF, reflux.
oxygen is probably not hydrogen bonded and is instead surrounded by three lipophilic amino acids.

The nitrogen atom of the thiazole ring is not involved in any discernible interaction with the binding site, whereas the methyl
group fits well into a pocket that is connected with the bottom end of the benzophenone pocket. This pocket is bounded by Cys-276, Met-330, Leu-344, and Met-355 residues, which are conserved in PPAR $\gamma$ and PPAR $\delta$ with the exception of Met330, which is valine in PPAR $\gamma$ and leucine in PPAR $\delta$, and Met355 , which is isoleucine in PPAR $\delta$. The smaller valine sidechain in PPAR $\gamma$ could accommodate slightly larger substituents at this position on the thiazole, while the branched isoleucine in PPAR $\delta$ might crowd larger substituents. In accordance with this hypothesis, replacement of the methyl by a trifluoromethyl group in $\mathbf{4 5}$ maintains PPAR $\alpha$ but increases PPAR $\gamma$ potency and decreases PPAR $\delta$ potency.

The phenyl-trifluoromethyl moiety makes a good fit into the lower pocket of the binding site and would be expected to contribute favorably to overall binding through van der Waals interactions. This observation is compatible with SAR data that show a relationship between size and lipophilicity of the parasubstituent and PPAR $\alpha$ potency. The decrease in potency seen with compounds bearing polar para-substituents ( $\mathbf{2 5} \mathbf{k}$ and $\mathbf{2 5 m}$ ) may be explained by the hydrophobic nature of the lower pocket. The inactivity of the methylsulfonyl analogue $\mathbf{2 5 1}$ may be due to a combination of polarity and steric constraints. metaSubstitution is well tolerated as evidenced by the fact that $\mathbf{2 5 i}$ is equipotent with 25a. Although the X-ray structure of 25a shows that one meta-position is directed toward Leu-247 and Ile-241, the other meta-position points into an open space and could easily accept a substituent. On the other hand, orthosubstitution would force the phenyl group out of plane with the thiazole and likely lead to a clash with the wall of the narrow pocket, a conjecture that is consistent with the decrease in potency seen with $\mathbf{2 5 j}$.

Pharmacokinetics in Rat and Dog (Table 6). Following intravenous administration of compound $\mathbf{2 5 a}(2.7 \mathrm{mg} / \mathrm{kg})$ to the rat, distribution to the tissues was limited with the volume of distribution ( $1 \mathrm{~L} / \mathrm{kg}$ ) similar to that of total body water ( 0.6 $\mathrm{L} / \mathrm{kg}$ ). Total plasma clearance was low ( $5 \mathrm{~mL} / \mathrm{min} / \mathrm{kg}$ ), representing about $6 \%$ of rat hepatic blood flow. The low clearance and moderate volume of distribution resulted in a plasma halflife of 2.4 h . Following a single oral dose of compound 25a at $3 \mathrm{mg} / \mathrm{kg}$, the maximum concentration of compound in the plasma was $1461 \mathrm{ng} / \mathrm{mL}$ after 1.5 h . The bioavailability was high ( $47 \%$ ).

Following intravenous administration of compound 25a to the dog at $2 \mathrm{mg} / \mathrm{kg}$, distribution to the tissues was limited with the volume of distribution $(2.8 \mathrm{~L} / \mathrm{kg})$ being greater than that of

Scheme $\mathbf{1 2}^{a}$

${ }^{a}$ Reagents and conditions: (a) ethyl-2-chloroacetoacetate, EtOH , reflux; (b) $\mathrm{MeSO}_{2} \mathrm{Cl}_{1} \mathrm{Et}_{3} \mathrm{~N}$, toluene/acetone; (c) LiOH (1.5 equiv), rt ; (d) 21, HOBT , EDC, DMF, $\mathrm{Et}_{3} \mathrm{~N}$, rt; (e) LiOH ( 1 equiv), THF/H2O, rt; (f) NaOH (1 N), EtOH, THF, $70^{\circ}$.

## Scheme $13{ }^{a}$


${ }^{a}$ Reagents and conditions: (a) $\mathrm{Br}_{2}$; (b) EtOH , reflux; (c) NaOH (1 N), EtOH, reflux; (d) 21, HOBT, EDC, DMF, $\mathrm{Et}_{3} \mathrm{~N}$, rt; (e) $\mathrm{NaOH}(1 \mathrm{~N}), \mathrm{THF}$, reflux.
total body water ( $0.6 \mathrm{~L} / \mathrm{kg}$ ). Total plasma clearance was moderate ( $13 \mathrm{~mL} / \mathrm{min} / \mathrm{kg}$ ), representing about $35 \%$ of hepatic blood flow in the dog. The moderate clearance and low volume of distribution resulted in a plasma half-life of 2.6 h . Following a single oral dose of compound $\mathbf{2 5 a}$ at $3 \mathrm{mg} / \mathrm{kg}$, the maximum concentration of compound in the plasma was $1449 \mathrm{ng} / \mathrm{mL}$. The bioavailability was high ( $85 \%$ ).

In Vivo Pharmacology. Several compounds in this series demonstrated profound in vivo activity in animal models of dyslipidemia, as illustrated by compound 25a. The human Apo-A-I-transgenic mouse model has been proposed to be potentially relevant to human disease, because in this model, fibrates give upregulation rather than the repression of Apo-A-I seen in other rodent models. ${ }^{19}$ Compound 25a shows similar PPAR $\alpha$ agonist potency and selectivity versus murine and human PPARs (murine PPAR $\mathrm{EC}_{50}, \alpha=15 \mathrm{nM} ; \delta=1000 \mathrm{nM} ; \gamma>10000$ $\mathrm{nM})$. When administered orally twice a day for 5 days, 25a, prepared as a suspension in $0.5 \%$ HPMC 100/1\% Tween80 at $\mathrm{pH}=7.0$, gave dose-related decreases in circulating TG,

## Scheme $14^{a}$


${ }^{a}$ Reagents and conditions: (a) ethyl-2-chloro-4,4,4-trifluoroacetoacetate, DMF, $100^{\circ} \mathrm{C}$; (b) $\mathrm{LiOH}(1 \mathrm{~N}), \mathrm{EtOH}$, rt; (c) $\mathrm{SOCl}_{2}, 21, \mathrm{NEt}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; (d) $\mathrm{NaOH}(1 \mathrm{~N})$, THF, reflux.

Table 2

|  |  |  |  |
| :---: | :---: | :---: | :---: |
|  | 1: $\mathrm{X}=\mathrm{S}$ | 14 |  |
| $10: \mathrm{X}=\mathrm{CH}_{2}$ |  |  |  |
|  | PPAR $\alpha^{a}$ | PPAR $\delta^{a}$ | PPAR $\gamma^{a}$ |
|  | $\mathrm{EC}_{50}{ }^{\text {a }}(\mu \mathrm{M}) ~, ~$ | $\mathrm{EC}_{50}{ }^{\text {b }}(\mu \mathrm{M}) ~(\%)$ | $\underset{\mathrm{EC}_{50}{ }^{\text {b }}(\mu \mathrm{M})}{ }$ |
| cmpd | (\% activation) ${ }^{\text {b,c }}$ | (\% activation) ${ }^{\text {b,c }}$ | (\% activation) ${ }^{\text {b,c }}$ |
| 1 | $1.888 \pm 0.1$ | $0.001 \pm 0.00009$ | $8.900 \pm 0.27$ |
|  | (108 $\pm 1 \%)$ | (97 $\pm 1 \%$ ) | (80 $\pm 1 \%$ ) |
| 10 | $2.0{ }^{\text {d }}$ | $0.004{ }^{\text {d }}$ | $10.0{ }^{\text {d }}$ |
|  | (85\%) | (117\%) | (74\%) |
| 14 | $0.21 \pm 0.08$ | $0.02 \pm 0.07$ | $1.87 \pm 0.79$ |
|  | (138 $\pm 61 \%)$ | (90 $\pm 10 \%$ ) | (76 $\pm 14.9 \%)$ |

${ }^{a}$ Data generated using cell based transient transfection assays described in ref 13. ${ }^{b} \pm$ SD. ${ }^{c}$ Percent of maximal activation of all compounds was compared to reference compounds normalized to $100 \%$. For the PPAR $\alpha$ activity, the reference compound was 25a. For the PPAR $\delta$ and PPAR $\gamma$ activities, the reference compounds were $\mathbf{1}$ and rosiglitazone, respectively. ${ }^{d}$ Results of a single experiment.

VLDLc, and LDLc and concomitant increases in HDLc (Table 7). The $E D_{50}$ for the HDL effect was approximately $1 \mathrm{mg} / \mathrm{kg}$. Circulating levels of Apo-A-I were also increased by the

Table 3

|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| cmpd | chain | PPAR $\alpha^{a}$ $\mathrm{EC}_{50}{ }^{b}(\mu \mathrm{M})$ (\% activation) ${ }^{b, c}$ | PPAR $\delta^{a}$ <br> $\mathrm{EC}_{50}{ }^{b}(\mu \mathrm{M})$ <br> (\% activation) ${ }^{b, c}$ | PPAR $\gamma^{a}$ $\mathrm{EC}_{50}(\mu \mathrm{M})$ $(\% \text { activation })^{b, c}$ |
| 14 | $-\mathrm{CH}_{2} \mathrm{CH}_{2}-$ | $\begin{aligned} & 0.21 \pm 0.08 \\ & (138 \pm 61 \%) \end{aligned}$ | $\begin{aligned} & 0.02 \pm 0.07 \\ & (90 \pm 10 \%) \end{aligned}$ | $\begin{aligned} & 1.87 \pm 0.79 \\ & (76 \pm 14.9 \%) \end{aligned}$ |
| 19 | - NHCO- | $\begin{aligned} & 9.5 \pm 0.5 \\ & (69 \pm 0 \%) \end{aligned}$ | $>10$ | > 10 |
| 23 | $-\mathrm{CH}_{2} \mathrm{NHCH}_{2}-$ | $\begin{aligned} & 0.07 \pm 0.01 \\ & (110 \pm 4 \%) \end{aligned}$ | $\begin{aligned} & 4.05 \pm 1.1 \\ & (61 \pm 2 \%) \end{aligned}$ | > 10 |
| 25a | $-\mathrm{CH}_{2} \mathrm{NHCO}-$ | $\begin{aligned} & 0.004 \pm 0.002 \\ & (95 \pm 14 \%) \end{aligned}$ | $\begin{aligned} & 2.83 \pm 1.18 \\ & (82 \pm 21 \%) \end{aligned}$ | > 10 |
| 27 | $-\mathrm{CH}_{2} \mathrm{NMeCO}-$ | $\begin{aligned} & 0.24^{d} \\ & (84 \%) \end{aligned}$ | $>10$ | > 10 |
| 34 | $-\mathrm{CONHCH}_{2}-$ | $\begin{aligned} & 1.32^{d} \\ & (109 \%) \end{aligned}$ | $\begin{aligned} & 6.26^{d} \\ & (62 \%) \end{aligned}$ | > 10 |
| 37 | - $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NHCO}-$ | $\begin{aligned} & 0.11 \pm 0.05 \\ & (129 \pm 58 \%) \end{aligned}$ | > 10 | > 10 |

${ }^{a}$ Data generated using cell based transient transfection assays described in ref $13 .{ }^{b} \pm$ SD. ${ }^{c}$ Percent of maximal activation of all compounds was compared to reference compounds normalized to $100 \%$. For the PPAR $\alpha$ activity, the reference compound was $\mathbf{2 5 a}$. For the PPAR $\delta$ and PPAR $\gamma$ activities, the reference compounds were $\mathbf{1}$ and rosiglitazone, respectively. ${ }^{d}$ Results of a single experiment.

Table 4

|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| cmpd | X | PPAR $\alpha^{a}$ <br> $\mathrm{EC}_{50}{ }^{b}(\mu \mathrm{M})$ (\% activation) $)^{b, c}$ | $\begin{gathered} \operatorname{PPAR} \delta^{a} \\ \mathrm{EC}_{50}(\mu \mathrm{M}) \\ (\% \text { activation })^{b, c} \end{gathered}$ | $\begin{gathered} \text { PPAR } \gamma^{a} \\ \mathrm{EC}_{50}{ }^{b}(\mu \mathrm{M}) \\ (\% \text { activation })^{b, c} \\ \hline \end{gathered}$ |
| 25a | $p-\mathrm{CF}_{3}$ | $\begin{aligned} & 0.004 \pm 0.002 \\ & (95 \pm 14 \%) \end{aligned}$ | $\begin{aligned} & 2.83 \pm 1.18 \\ & (82 \pm 21 \%) \end{aligned}$ | > 10 |
| 25b | $p$-t-butyl | $\begin{aligned} & 0.005 \pm 0.000 \\ & (66 \pm 6 \%) \end{aligned}$ | > 10 | $\begin{aligned} & 0.86 \pm 0.01 \\ & (78 \pm 0 \%) \end{aligned}$ |
| 25c | $p-\mathrm{OCF}_{3}$ | $\begin{aligned} & 0.004 \pm 0.000 \\ & (116 \pm 5 \%) \end{aligned}$ | $\begin{aligned} & 4.6 \pm 0.5 \\ & (73 \pm 3 \%) \end{aligned}$ | > 10 |
| 25d | $p-\mathrm{Cl}$ | $\begin{aligned} & 0.01 \pm 0.003 \\ & (94 \pm 7 \%) \end{aligned}$ | $\begin{aligned} & 5.4 \pm 0.9 \\ & (85 \pm 9 \%) \end{aligned}$ | > 10 |
| 25e | $p-\mathrm{NO}_{2}$ | $\begin{aligned} & 0.03 \pm 0.01 \\ & (111 \pm 17 \%) \end{aligned}$ | >10 | > 10 |
| $25 f$ | $p$-OMe | $\begin{aligned} & 0.01 \pm 0.01 \\ & (103 \pm 8 \%) \end{aligned}$ | > 10 | > 10 |
| 25g | $p-\mathrm{F}$ | $\begin{aligned} & 0.04^{d} \\ & (118 \%) \end{aligned}$ | > 10 | > 10 |
| 25h | H | $\begin{aligned} & 0.12 \pm 0.02 \\ & (107 \pm 5 \%) \end{aligned}$ | > 10 | > 10 |
| $25 i$ | $m-\mathrm{CF}_{3}$ | $\begin{aligned} & 0.004^{d} \\ & (106 \%) \end{aligned}$ | >25 | >25 |
| 25j | $o-\mathrm{CF}_{3}$ | $\begin{aligned} & 0.465^{d} \\ & (82 \%) \end{aligned}$ | $\begin{aligned} & 0.808^{d} \\ & (74 \%) \end{aligned}$ | >25 |
| 25k | $p-\mathrm{NH}_{2}$ | $\begin{aligned} & 5.64 \pm 0.29 \\ & (67 \pm 4 \%) \end{aligned}$ | > 10 | > 10 |
| 251 | $p-\mathrm{OSO}_{2} \mathrm{Me}$ | $>10$ | $>10$ | $>10$ |
| 25m | $p$-OH | $>10$ | $>10$ | > 10 |

${ }^{a}$ Data generated using cell based transient transfection assays described in ref. 13. ${ }^{b} \pm$ SD. ${ }^{c}$ Percent of maximal activation of all compounds was compared to reference compounds normalized to $100 \%$. For the PPAR $\alpha$ activity, the reference compound was 25a. For the PPAR $\delta$ and PPAR $\gamma$ activities, the reference compounds were $\mathbf{1}$ and rosiglitazone, respectively. ${ }^{d}$ Results of a single experiment.
treatment (data not shown), consistent with the mechanism of action. As a reference, Fenofibrate was tested in human Apo-A-I transgenic mice at $50 \mathrm{mg} / \mathrm{kg}$. It produced the expected profile with a decrease in plasma TG $(-43 \%)$, VLDL cholesterol ( $-64 \%$ ), and LDL cholesterol ( $-78 \%$ ), as well as an increase

Table 5

|  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| \# | Heterocycle | PPAR $\alpha^{\text {a }}$ | PPAR $\delta^{\text {a }}$ | PPAR $\gamma^{\text {a }}$ |
|  |  | $\mathrm{EC}_{50}(\mu \mathrm{M})^{\text {b }}$ | $\mathrm{EC}_{50}(\mu \mathrm{M})^{\mathrm{b}}$ | $\mathrm{EC}_{50}(\mu \mathrm{M})^{\mathrm{b}}$ |
|  |  | (\%activation) ${ }^{\text {b,c }}$ | (\%activation) ${ }^{\text {b,c }}$ | (\%activation) ${ }^{\text {b,c }}$ |
| 25a |  | $0.004 \pm 0.002$ | $2.83 \pm 1.18$ | >10 |
|  |  | ( $95 \pm 14 \%$ ) | (82 $21 \%$ ) |  |
| 41 |  | $1.89{ }^{\text {d }}$ | $0.66{ }^{\text {d }}$ | >10 |
|  |  | (171\%) | (103\%) |  |
| 45 |  | $0.004 \pm 0.002$ | $8.24 \pm 1.75$ | $0.73 \pm 0.12$ |
|  |  | (98 $\pm 8 \%$ ) | (54 $\pm 5 \%$ ) | (63 $\pm 1 \%$ ) |

${ }^{a}$ Data generated using cell based transient transfection assays described in ref 13. ${ }^{b} \pm$ SD. ${ }^{c}$ Percent of maximal activation of all compounds was compared to reference compounds normalized to $100 \%$. For the PPAR $\alpha$ activity, the reference compound was 25a. For the PPAR $\delta$ and PPAR $\gamma$ activities, the reference compounds were $\mathbf{1}$ and rosiglitazone, respectively. ${ }^{d}$ Results of a single experiment.


Figure 1. X-ray crystal structure of 25a complexed with the PPAR $\alpha$ ligand binding domain. The molecular surface of the binding site is represented by blue dots. Interactions of the head group amino acids with the carboxylate moiety of the ligand are shown as white dotted lines. The hydrogens shown here and in Figure 2a,b were not visible in the electron density and were instead positioned to optimize their hydrogen bonding interactions.
in HDL cholesterol ( $+26 \%$ ). The finding that $\mathbf{2 5 a}$ is able to lower LDLc and TG and increase HDL cholesterol in the Apo-A-I-transgenic mouse model suggests that it will deliver significant therapeutic benefit in the treatment of dyslipidemia and hypertriglyceridemia. Compound 25a (GW590735) has been progressed to clinical trials for the treatment of diseases of lipid imbalance.

## Conclusions

Starting from the selective PPAR $\delta$ agonist 1, which showed weak potency on PPAR $\alpha$, we have developed a new series of highly potent and selective PPAR $\alpha$ agonists, one of which, 25a, has progressed to clinical evaluation. The superior potency and PPAR subtype selectivity of $\mathbf{2 5}$ a suggest that this compound offers the potential to deliver significantly improved therapeutic benefits over the fibrates in dyslipidemia and hypertriglyceridemia.


Figure 2. (a) Environment of the linking amide group of 25a: hydrogen bond network surrounding the carbonyl. Water molecules are shown as white spheres; hydrogen bond interactions are represented as green dotted lines; distances are indicated in $\AA$ and represented as yellow dotted lines; for clarity of the figure, the amino acid backbone was deleted for Ser-280 and Thr-279. (b) Environment of the linking amide group of 25a: triad of sulfur containing amino acids surrounding the amide NH . $\mathrm{NH}-\mathrm{S}$ distances are indicated in $\AA$ and represented as yellow dotted lines.

Table 6. Pharmacokinetic Data for Compound 25a

| PK parameter | rat | dog |
| :--- | :---: | :---: |
| $\mathrm{Cl}(\mathrm{mL} / \mathrm{min} / \mathrm{kg})$ | 5 | 13 |
| $\mathrm{Vd}(\mathrm{L} / \mathrm{kg})$ | 1 | 2.8 |
| $T_{1 / 2}(\mathrm{~h})$ | 2.4 | 2.6 |
| $F(\%)$ | 47 | 85 |

Table 7. Effect of 25a in the Human Apo-A-I Transgenic Mouse Model ${ }^{a}$

| oral dose <br> $(\mathrm{mg} / \mathrm{kg}$, b.i.d. $)$ | VLDL chol <br> $(\mathrm{g} / \mathrm{L}) \%$ | LDL chol <br> $(\mathrm{g} / \mathrm{L}) \%$ | HDL chol <br> $(\mathrm{g} / \mathrm{L}) \%$ | TG <br> $(\mathrm{g} / \mathrm{L}) \%$ |
| :---: | :---: | :---: | :---: | :---: |
| 0.5 | $-48 \pm 8$ | $-61 \pm 2$ | $+43 \pm 4$ | $-37 \pm 6$ |
| 1.0 | $-63 \pm 10$ | $-65 \pm 3$ | $+51 \pm 19$ | $-48 \pm 11$ |
| 5.0 | $-86 \pm 5$ | $-69 \pm 5$ | $+86 \pm 18$ | $-65 \pm 6$ |

${ }^{a}$ All changes are statistically significant ( $p<0.05$ ), as determined by one-way ANOVA analysis.

## Experimental Section

General Methods. All commercial chemicals and solvents are reagent grade and were used without further purification unless otherwise specified. All reactions except those in aqueous media were carried out with the use of standard techniques for the exclusion of moisture. Reactions were monitored by thin-layer chromatography on 0.25 mm silica gel plates ( $60 \mathrm{~F}-254$, E. Merck) and visualized with UV light or 5\% phosphomolybdic acid in 95\% ethanol.

Final compounds were typically purified either by flash chromatography on silica gel (E. Merck, $40-63 \mathrm{~mm}$ ) or by recrystallization. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on either a Brucker 300 MHz Avance DPX or a Bruker 400 MHz Avance DRX instrument. Chemical shifts are reported in parts per million (ppm, d units).

Splitting patterns are designed as s , singlet; d , doublet; t , triplet; q , quartet; m, multiplet; bs, broad singlet. Low-resolution mass spectra (MS) were recorded on a Micromass platform LC or a Agilent GC/ MS spectrometer. High-resolution mass spectra were recorded on a Micromass LCT (TOF) spectrometer. Mass spectra were acquired in either positive or negative ion mode under electrospray ionization (ESI) or atmospheric pressure chemical ionization (APCI) methods.

Analytical high performance liquid chromatography (HPLC; system A) was conducted on a Chromolith Performance RP 18 column ( $100 \times 4.6 \mathrm{~mm}$ id) eluting with 0.01 M ammonium acetate in water and $100 \%$ acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$, using the following elution gradient: 0 to $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 4 min and $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 1 min at $5 \mathrm{~mL} / \mathrm{min}$ at a temperature of $30^{\circ} \mathrm{C}$.

Analytical HPLC (system B) was conducted on a XTERRAMS C18 column ( $30 \times 3 \mathrm{~mm}$ id, $2.5 \mu \mathrm{~m}$ ) eluting with 0.01 M ammonium acetate in water and $100 \%$ acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$, using the following elution gradient: 0 to $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 4 min and $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 1 min at $1.1 \mathrm{~mL} / \mathrm{min}$ at a temperature of 40 ${ }^{\circ} \mathrm{C}$.

Analytical HPLC (system C) was conducted on a Uptispherehsc column ( $3 \mu \mathrm{~m} 33 \times 3 \mathrm{~mm}$ id) eluting with 0.01 M ammonium acetate in water and $100 \%$ acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$ using the following elution gradient: $0-5 \% \mathrm{CH}_{3} \mathrm{CN}$ over 0.5 min , gradient $5-100 \%$ $\mathrm{CH}_{3} \mathrm{CN}$ over 3.25 min , and $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 0.75 min at 1.3 $\mathrm{mL} / \mathrm{mn}$ at a temperature of $40^{\circ} \mathrm{C}$.

Analytical HPLC (system D) was conducted on a Uptispherehsc column ( $3 \mu \mathrm{~m} 33 \times 3 \mathrm{~mm}$ id) eluting with 0.01 M ammonium acetate in water and $100 \%$ acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$ using the following elution gradient: $0-5 \% \mathrm{CH}_{3} \mathrm{CN}$ over 0.5 min , gradient $5-100 \%$ $\mathrm{CH}_{3} \mathrm{CN}$ over 6.25 min , and $100 \% \mathrm{CH}_{3} \mathrm{CN}$ over 0.75 min at 1.3 $\mathrm{mL} / \mathrm{mn}$ at a temperature of $40^{\circ} \mathrm{C}$.

The purity of key compounds was determined on two analytic HPLC systems using UV detection.

Pharmacokinetics in Rat and Dog. Compound 25a was administered to Wistar rats $(n=15)$ by oral gavage at dose of 3 $\mathrm{mg} / \mathrm{kg}$ in pH 7 buffer, $0.1 \%$ Tween 80 and by intravenous injection via the penis vein $(n=30)$ at a dose of $2.7 \mathrm{mg} / \mathrm{kg}$ in $10 \%$ DMSO and PEG200. Compound 25a was administered orally to male beagle dogs $(n=3)$ by stomach intubation at a dose of $3 \mathrm{mg} / \mathrm{kg}$ in pH 7 buffer, $0.1 \%$ Tween 80 and by intravenous injection via the cephalic vein at a dose of $2 \mathrm{mg} / \mathrm{kg}$ in $10 \%$ DMSO and PEG200. Blood samples were placed on wet ice, and plasma was collected after centrifugation. Plasma samples were stored frozen at $-20^{\circ} \mathrm{C}$ until time of analysis. Plasma samples ( 0.5 mL ) were diluted with 1:1 buffer $\left(\mathrm{NaH}_{2} \mathrm{PO}_{4}, 0.1 \mathrm{M}, \mathrm{pH} 4\right)$ and then extracted with ethyl acetate ( 5 mL ). The ethyl acetate was evaporated, and the residue was resuspended in $200 \mu \mathrm{~L}$ of mobile phase (water/acetonitrile/ TFA; 30v/70v/0.1\%). Samples were analyzed by high-performance liquid chromatography spectrometric analysis (LC/MS/MS). Pharmacokinetic parameters were determined by SIPHAR.

Apo-A-I Transgenic Mouse Model. Male C57BL/6 mice transgenic for human ApoA-I were obtained from Charles River Laboratories (L'Arbresle, France) and randomized into treatment groups of $n=5$ animals. Twice a day oral administration of vehicle ( $0.5 \%$ HPMC $/ 1 \%$ Tween80, $\mathrm{pH}=7.0$ ) or indicated doses of compound as a suspension began when animals were nine weeks old and lasted for 5 days. Animals were fasted overnight before blood samples were taken by intracardiac puncture. Whole liver was collected and weighed. Blood samples were left for 30 min at $37^{\circ} \mathrm{C}$ to coagulate and centrifuged 10 min at 10000 rpm . Total serum fraction was then collected and frozen at $-20^{\circ} \mathrm{C}$ until use. Total cholesterol and total TG were dosed using kits 61219 and 61236, respectively, following manufacturer instructions (Bio Mérieux, Marcy-l'étoile, France). After 10 min of incubation at 37 ${ }^{\circ} \mathrm{C}$, the colorimetric reaction was read at 492 nm with an iEMS reader (ThermoLife Sciences, Cergy-Pontoise, France). Cholesterol HDL, LDL, and VLDL fractions were separated by HPLC. Samples were diluted $1 / 5$ in phosphate buffer $\left(\mathrm{Ca}^{++}\right.$and $\mathrm{Mg}^{-}$free) and filtered on $0.45 \mu \mathrm{~m}$ to remove excess proteins before HPLC. All changes reported with an asterisk are statistically significant ( $p<$ 0.05 ) as determined by one-way ANOVA analysis.

Chemicals. Substituted Thiobenzamides 2a-j. Substituted thiobenzamides were either purchased from commercial suppliers (2a, 2d, 2f, 2h Lancaster; 2c Fluorochem; 2g Acros) or were prepared using one of the routes $\mathbf{a}$ or $\mathbf{b}$ described below.

Procedure a for the Preparation of $\mathbf{2 b}$. To a solution of $\mathrm{P}_{4} \mathrm{~S}_{10}$ $(0.2 \mathrm{mmol})$ in toluene ( 100 mL ) was added $\mathrm{NaHCO}_{3}(2 \mathrm{mmol})$, and the mixture was heated to reflux for about 30 min . The substituted benzamide ( 1 mmol ) was added, and the reaction was stirred at $90^{\circ} \mathrm{C}$ for 1 h . The reaction was then evaporated to dryness, treated with brine ( 100 mL ), and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 50$ mL ). The organic phase was dried, filtered, and evaporated to afford the final product. In this manner were obtained products $\mathbf{2 b}$ (orange solid; $49 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 7.7$ (d, 2H), 7.4 (bs, 1H), 7.3 (d, $2 \mathrm{H}), 7.0(\mathrm{bs}, 1 \mathrm{H}), 1.2(\mathrm{~s}, 9 \mathrm{H})$.

Procedure b for the Preparation of Substituted Thiobenzamides $2 \mathbf{e}, \mathbf{2 i}$, and $\mathbf{2 j}$. To the substituted benzonitrile ( 1 mmol ) in DMF ( 30 mL ) was added dropwise DMF ( 21 mL ) saturated with $\mathrm{HCl}_{(\mathrm{g})}$ over a period of 1 min . Thioacetamide ( 2 mmol ) was then added, and the reaction was heated to $100{ }^{\circ} \mathrm{C}$ for $1 \mathrm{~h} . \mathrm{HCl}_{(\mathrm{g})}$ was bubbled in for about 1 min , and stirring was continued at $100{ }^{\circ} \mathrm{C}$ for another 18 h . The reaction was cooled to rt, treated with ice, and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 250 \mathrm{~mL})$. The organic phase was washed with $\mathrm{H}_{2} \mathrm{O}(3 \times 300 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated to dryness. The residue was washed with a mixture of isopropyl ether/pentane (1:3) to afford the final product. In this manner were obtained products $\mathbf{2 e}$ (orange solid; $83 \%$ ), ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ) $\delta 10.1(\mathrm{bs}, 1 \mathrm{H}), 9.7(\mathrm{bs}, 1 \mathrm{H}), 8.1(\mathrm{~d}, 2 \mathrm{H}), 7.9(\mathrm{~d}, 2 \mathrm{H})$; $2 \mathbf{i}$ (light yellow solid; 88\%), ${ }^{1} \mathrm{H}$ NMR (DMSO) $\delta 8.20$ (bs, 1H), 8.08 (s, 1H), 8.02 (d, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.91(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H})$, 7.76 (t, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.34(\mathrm{bs}, 1 \mathrm{H})$; and $\mathbf{2 j}$ (light yellow solid; $47 \%$ ), ${ }^{1} \mathrm{H}$ NMR (DMSO) $\delta 7.9-7.4(\mathrm{~m}, 4 \mathrm{H}), 7.8(\mathrm{bs}, 1 \mathrm{H}), 7.06$ (bs, 1H).

Ethyl 4-Methyl-2-[4-(trifluoromethyl)phenyl]-1,3-thiazole-5carboxylate (3a). To a suspension of of 4-(trifluoromethyl)thiobenzamide 2 a ( $302.4 \mathrm{~g}, 1.47 \mathrm{~mol}$.) in ethyl alcohol ( $1.5 \mathrm{~L}, 5 \mathrm{vol}$ ) was added at room temperature ethyl 2-chloroacetoacetate (203.8 $\mathrm{mL}, 1$ equiv). The solution was refluxed for 24 h , and then the solvent was removed under reduced pressure. The solid material was stirred with cooled hexane ( 500 mL ) for 30 min , filtered, and washed with hexane ( $2 \times 150 \mathrm{~mL}$ ). Drying gave crude compound $3 \mathrm{a}(352.9 \mathrm{~g})$. A second crop of 25.7 g was obtained by concentration of the hexane filtrates to 50 mL , giving an overall yield of 378.6 $\mathrm{g}(81.5 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 8.06(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.69$ (d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.78(\mathrm{~s}, 3 \mathrm{H}), 1.39(\mathrm{t}, J=7.1$ $\mathrm{Hz}, 3 \mathrm{H}$ ).

4-Methyl-2-[4-(trifluoromethyl)phenyl]-1,3-thiazole-5-carboxylic Acid (17a). To a cooled solution of $\mathbf{3 a}(378.6 \mathrm{~g}, 1.2 \mathrm{~mol})$ in ethyl alcohol ( $2 \mathrm{~L}, 5 \mathrm{vol}$ ) was added a solution of sodium hydroxide ( $96.15 \mathrm{~g}, 2$ equiv) in 2 L of water. The solution was heated at $85^{\circ} \mathrm{C}$ for 1.5 h . After evaporation of the ethyl alcohol, the aqueous solution was diluted with 2 L of water and acidified to $\mathrm{pH}=1$ with concentrated aqueous hydrochloric acid. The solid material was filtered and washed twice with 1 L of water and 1 L of dichloromethane. After drying in a vacuum oven, 17a ( 267.2 g ) was obtained as an off-white powder. A second crop of 25.7 g was obtained by concentration of the dichloromethane and triturating with pentane, giving an overall yield of 17 a of $292.9 \mathrm{~g}(85 \%) .{ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ): $\delta 13.51$ (bs, 1H), 8.15 (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.84 (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.68 ( $\mathrm{s}, 3 \mathrm{H}$ ).

2-Methyl-2-[4-\{[(4-methyl-2-[4-trifluoromethylphenyl]thiazol-5-ylcarbonyl)amino]methyl\}phenoxy]propionic Acid Ethyl Ester (24a). A suspension of crude $\mathbf{1 7 a}(38.7 \mathrm{~g}, 0.13 \mathrm{~mol})$ in thionyl chloride ( $200 \mathrm{~mL}, 5 \mathrm{vol}$ ) was refluxed for 3 h . After return to room temperature, the thionyl chloride was removed under reduced pressure, and the residue was washed twice with toluene ( 100 mL ) and evaporated to dryness. The crude acid chloride obtained (offwhite solid) was used without purification. To a mixture of crude 21 ( 35.5 g ; 1 equiv/LC-MS purity: $90 \%$ ) and triethylamine ( 20.62 $\mathrm{mL}, 1.1$ equiv) in dichloromethane ( $350 \mathrm{~mL}, 10 \mathrm{vol}$ ), maintained at $10^{\circ} \mathrm{C}$, was added portionwise the acid chloride over 20 min , and the mixture was stirred at room temperature overnight. The
reaction was quenched by addition of water ( 200 mL ) and was stirred for 5 min . The aqueous layer was extracted with dichloromethane ( $2 \times 200 \mathrm{~mL}$ ). The combined organic layers were washed with hydrochloric acid ( $1 \mathrm{~N}, 200 \mathrm{~mL}$ ), water ( 200 mL ), saturated aqueous sodium carbonate ( 200 mL ), and brine ( 200 mL ). After drying over magnesium sulfate, filtration, and concentration to dryness, the crude material was suspended in isopropyl ether (200 mL ), triturated, filtered, and dried to give 24a ( $47.6 \mathrm{~g}, 69.7 \%$ ) as a white powder: ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ): $\delta 8.87(\mathrm{t}, J=5.6 \mathrm{~Hz}$, $1 \mathrm{H}), 8.14$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.87 (d, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.23$ (d, $J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.75(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 4.37(\mathrm{~d}, J=5.8 \mathrm{~Hz}$, $2 \mathrm{H}), 4.15(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.63(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 6 \mathrm{H}), 1.16(\mathrm{t}$, $J=7.1 \mathrm{~Hz}, 3 \mathrm{H})$.

2-Methyl-2-[4-\{[(4-methyl-2-[4-trifluoromethylphenyl]-thia-zol-5-ylcarbonyl)amino]methyl $\}$ phenoxy $]$ propionic Acid (25a). To a solution of 24a ( $230.8 \mathrm{~g}, 0.46 \mathrm{~mol}$ ) in $1.2 \mathrm{~L}(5 \mathrm{vol})$ of tetrahydrofuran was added aqueous sodium hydroxide ( $1 \mathrm{~N}, 480$ $\mathrm{mL}, 1.05$ equiv). The solution was stirred under reflux for 18 h . After removal of THF under reduced pressure, 1 N NaOH ( 500 $\mathrm{mL})$ and methyl alcohol ( 100 mL ) were added. The aqueous layer was extracted with dichloromethane $(2 \times 400 \mathrm{~mL})$ and acidified to $\mathrm{pH}=1$ with concentrated aqueous hydrochloric acid. The oily residue was extracted with dichloromethane ( $3 \times 400 \mathrm{~mL}$ ), and the combined organic layers were washed with brine ( 600 mL ). After drying over magnesium sulfate, filtration, and concentration to dryness, the oily residue was triturated with pentane ( 500 mL ), filtered, and washed with pentane ( $2 \times 250 \mathrm{~mL}$ ) to give, after drying, crude compound 25a (207.2 g) as a white powder. The solid material was dissolved in $310 \mathrm{~mL}(1.5 \mathrm{vol})$ of boiling toluene. After filtration of the hot solution and return to room temperature, the crystallized material was filtered, washed with toluene ( $2 \times$ 200 mL ), and dried in vacuo to give $\mathbf{2 5 a}(196.3 \mathrm{~g}, 90 \%)$, as a white powder; mp 130-131 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ) $\delta 13.0$ (bs, 1 H ), $8.87(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.87(\mathrm{~d}, J=8.3$ $\mathrm{Hz}, 2 \mathrm{H}), 7.22(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.78(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 4.37$ (d, $J=5.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), $2.63(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}(\mathrm{APCI}) \mathrm{m} / \mathrm{z}$ $479(\mathrm{M}+\mathrm{H})^{+}$. HRMS calcd for $\mathrm{C}_{23} \mathrm{H}_{21} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}(\mathrm{M}+\mathrm{H})$, 479.1252; found, 479.1236. Analytical HPLC $t_{\mathrm{R}}=1.85 \mathrm{~min}, 100 \%$ pure $(\mathrm{A}) ; t_{\mathrm{R}}=2.16 \mathrm{~min}, 100 \%$ pure (C).

2-Methyl-2-\{[4-(\{[(4-methyl-2-\{4-[(methylsulfonyl)oxy]phen-yl\}-1,3-thiazol-5-yl)carbonyl]amino\}methyl)phenyl]oxy\}propanoic Acid (25I). To a solution of $241(0.34 \mathrm{~g}, 0.64 \mathrm{mmol})$ in THF (10 mL ) was added a solution of LiOH in water ( 1 equiv, 2.5 mL ), and the mixture was stirred at rt for 24 h . The THF was evaporated, and the aqueous phase was acidified with $\mathrm{HCl}(1 \mathrm{~N})$ and then extracted with ethyl acetate ( 50 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated to dryness. The residue was chromatographed, eluting with $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95 / 5$ then $90 / 10$ to afford $\mathbf{2 5 1}$ as a white solid ( $50 \mathrm{mg}, 15 \%$ ). ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ) $\delta$ $8.78(\mathrm{t}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.03(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.48(\mathrm{~d}, J=8.7$ $\mathrm{Hz}, 2 \mathrm{H}), 7.13(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.80(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.33$ $(\mathrm{d}, J=5.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.44(\mathrm{~s}, 3 \mathrm{H}), 2.61(\mathrm{~s}, 3 \mathrm{H}), 1.40(\mathrm{~s}, 6 \mathrm{H}) . \mathrm{MS}$ (APCI) $m / z 505(\mathrm{M}+\mathrm{H})^{+}$. HRMS calcd for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{~S}_{2}(\mathrm{M}+$ H), 505.1103; found, 505.1088. Analytical HPLC $t_{\mathrm{R}}=1.59 \mathrm{~min}$, $100 \%$ pure (A); $t_{\mathrm{R}}=1.93 \mathrm{~min}, 100 \%$ pure (C).

2-(\{4-[(\{[2-(4-Hydroxyphenyl)-4-methyl-1,3-thiazol-5-yl]-carbonyl\}amino)methyl]phenyl\}oxy)-2-methylpropanoic Acid ( $\mathbf{2 5 m}$ ). To a solution of $\mathbf{2 4 1}(0.20 \mathrm{~g}, 0.38 \mathrm{mmol})$ in THF ( 30 mL ) and $\mathrm{EtOH}(5 \mathrm{~mL}$ ) was added NaOH solution ( $1 \mathrm{~N}, 1.85 \mathrm{~mL}, 5$ equiv), and the mixture was stirred at $70^{\circ} \mathrm{C}$ for 4 h . The mixture was cooled to room temperature, and the solution was acidified with $\mathrm{HCl}(1 \mathrm{~N})$ and extracted with ethyl acetate $(3 \times 50 \mathrm{~mL})$. The combined organic layers were washed with $\mathrm{H}_{2} \mathrm{O}$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated to dryness. The residue was crystalized in a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and THF to afford $\mathbf{2 5 m}(0.12 \mathrm{~g}, 74 \%)$ as white crystals; mp 151-152 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ): $\delta 13.00(\mathrm{~s}, 1 \mathrm{H})$, $10.13(\mathrm{~s}, 1 \mathrm{H}), 8.67(\mathrm{~s}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.21(\mathrm{~d}, J=$ $8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.86(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.77(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H})$, 4.33 (d, $J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.57$ (s, 3H), 1.48 ( $\mathrm{s}, 6 \mathrm{H}$ ). MS (APCI) $m / z 427(M+H)^{+}$. HRMS calcd for $\mathrm{C}_{22} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}(\mathrm{M}+\mathrm{H})$,

Table 8. Data Collection and Refinement Statistics

| crystallization | PPAR $\alpha / \mathrm{SRC} 1$ peptide 25a |
| :---: | :---: |
| PDB code | to be deposited |
| unit cell dimensions |  |
| $a(\AA)$ | 61.3 |
| $b(\AA)$ | 103.5 |
| $c(\AA)$ | 49.9 |
| space group | $P 2{ }_{1} 2_{1} 2$ |
| molecules per asymmetric unit | 1 |
| resolution ( A ) | 50-1.8 |
| number of unique reflections | 30145 |
| data redundancy | 7.1 |
| completeness (\%) | 99.4 |
| $R_{\text {symm }}$ (\%) | 4.5 |
| $\langle I / \sigma\rangle$ | 39.4 |
| Refinement |  |
| $R$ | 20.4 |
| $R_{\text {free }}$ | 22.7 |
| r.m.s. deviation from ideality |  |
| bond length ( A ) | 0.005 |
| bond angle (degrees) | 1.288 |
| average B-factor ( $\AA^{2}$ ) |  |
| all atoms | 27.0 |
| protein atoms | 25.9 |
| water molecules | 37.5 |
| ligand | 19.7 |

427.1328; found, 427.1319. Analytical HPLC $t_{\mathrm{R}}=2.06 \mathrm{~min}, 100 \%$ pure (B); $t_{\mathrm{R}}=1.78 \mathrm{~min}, 98.9 \%$ pure (C).

Ethyl 5-Methyl-2-[4-(trifluoromethyl)phenyl]-1,3-thiazole-4carboxylate (38). To a solution of 2-ketobutyric acid (5.0 g, 49 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL}$ ) was slowly added bromine ( 1 equiv; 2.5 mL ), and the mixture was stirred at rt for 15 min . The solvent was evaporated. Toluene was added and the solvent was evaporated again. The residue was dissolved in EtOH, then 1a $8.5 \mathrm{~g} ; 0.85$ equiv) was added, and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 18 h . The solvent was evaporated, and the crude product was chromatographed, eluting with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to afford $\mathbf{3 8}(4.56 \mathrm{~g}, 35 \%)$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 8.04(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.45(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.82(\mathrm{~s}, 3 \mathrm{H}), 1.43(\mathrm{t}, J=$ 7.2 Hz, 3H).

Protein Preparation. The PPAR $\alpha$ ligand binding domain (amino acids 192-468) with an N-terminal 6xHis tag was expressed using the T7 promoter of plasmid vector pRSETA. BL21(DE3) E. coli cells transformed with this expression vector were grown at $24^{\circ} \mathrm{C}$ in shaker flasks for 66 h . The cells were harvested, resuspended, and lysed. The lysed cells were centrifuged, and the supernatant was loaded on a Ni -agarose column. The column was washed with 150 mL of buffer A ( $10 \%$ glycerol, 20 mM HEPES $\mathrm{pH} 7.5,25$ mM imidazole), and the protein was eluted with a 450 mL gradient of buffer B ( $10 \%$ glycerol, 20 mM HEPES $\mathrm{pH} 7.5,500 \mathrm{mM}$ imidazole). The protein, which eluted at $20 \%$ buffer B, was diluted with one volume of buffer C ( 20 mM HEPES, $\mathrm{pH} 7.5,1 \mathrm{mM}$ EDTA) and loaded on a 100 mL S-Sepharose (Pharmacia, Peapack, New Jersey) column. The column was washed with 100 mL buffer C, and the PPAR $\alpha$ LBD protein was eluted with a 200 mL gradient of buffer D ( 20 mM HEPES, $\mathrm{pH} 7.5,10 \mathrm{mM}$ DTT, 1 M ammonium acetate). The PPAR $\alpha$ LBD eluted from the column at $43 \%$ buffer D. The protein yield was $9 \mathrm{mg} / \mathrm{L}$ of cells grown and was $>95 \%$ pure, as determined by SDS-PAGE analysis.

The protein was then diluted to $1 \mathrm{mg} / \mathrm{mL}$ with buffer C such that the final buffer composition was 220 mM ammonium acetate, 20 mM HEPES $\mathrm{pH} 7.5,1 \mathrm{mM}$ EDTA, and 1 mM DTT. The peptide SRC1 ${ }^{16}$ was added in a molar ratio of 1.5 as a $2 \mathrm{mg} / 100 \mu$ L DMSO stock. The ligand was then added in a 5:1 molar ratio as a $2 \mathrm{mg} /$ $100 \mu \mathrm{~L}$ DMSO stock and spun at 4 K for 20 min to clarify the solution before concentrating in Centriprep 10 filtration units (Millipore, Bedford, Massachusetts). The solution containing the PPAR $\alpha$ LBD-SRC1 complexes was concentrated to approximately $10 \mathrm{mg} / \mathrm{mL}$ with $80 \%$ yield.

Crystallization and Data Collection. The crystals were grown at room temperature using the hanging drop vapor diffusion method.

The hanging drops contained $1 \mu \mathrm{~L}$ of the above protein-ligand solution and $1 \mu \mathrm{~L}$ of well buffer comprising 7\% PEG 3350, 200 mM NaF , and $12 \%$ 2,5-hexanediol. Crystals formed in the space group $P 2_{1} 2_{1} 2$ with cell dimensions $a=61.3 \AA, b=103.5 \AA$, and $c=49.9$ A. Each asymmetric unit contained a single LBD complex with $45 \%$ solvent content. X-ray data was collected at the IMCA 17-ID line (Argonne, IL) and processed with HKL2000. ${ }^{20}$ Statistics are summarized in Table 8.

Structure Determination and Refinement. The structure was determined by molecular replacement methods with the program AmoRe, ${ }^{21}$ as implemented in the CCP4 suite. ${ }^{22}$ The structure of the PPAR $\alpha$ LBD, ${ }^{23}$ residues $167-441$, was used as the initial model. The best fitting solution gave a correlation coefficient of $70 \%$ and an R-factor of $33 \%$. Model building was performed with the software program QUANTA, and structure refinement was carried out with the CNS software program. ${ }^{24}$ Refinement statistics are summarized in Table 8.

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Supporting Information Available: Experimental details and data for compounds $\mathbf{3 b}-\mathbf{k}, 4,5,6,7,8,9,10,11,12,13,14,15$, 16, 17b-k, 18, 19, 20, 21, 22, 23, 24b-j, 25b-j, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 24k, 25k, 24l, 39, 40, 41, 42, 43, 44, and 45 plus a table of analytical data for target compounds. This material is available free of charge via the Internet at http:// pubs.acs.org

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[^1]:    ${ }^{a}$ Reagents and conditions: (a) ethyl-2-chloroacetoacetate, EtOH , reflux; (b) $\mathrm{LiAlH}_{4}, \mathrm{THF}, 0{ }^{\circ} \mathrm{C}$; (c) $\mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt; (d) conc. HCl , petroleum ether, $37 \% \mathrm{HCHO} / \mathrm{H}_{2} \mathrm{O}$; (e) $\mathrm{PPh}_{3}$, toluene, reflux; (f) $\mathrm{NaH}\left(60 \%\right.$ in mineral oil), anhydrous EtOH , rt; (g) $10 \% \mathrm{Pd} / \mathrm{C}, \mathrm{EtOH} /$ dioxan, $\mathrm{H}_{2}$; (h) $\mathrm{NaOH}(1 \mathrm{~N})$, dioxane,

